Europäisches Patentamt
European Patent Office
Office européen des brevets



(11) EP 1 591 417 A1

(12)

EUROPEAN PATENT APPLICATION published in accordance with Art. 158(3) EPC

- (43) Date of publication: 02.11.2005 Bulletin 2005/44
- (21) Application number: 03814527.2
- (22) Date of filing: 01.04.2003

- (51) Int Cl.7: C01B 31/02
- (86) International application number: PCT/JP2003/004182
- (87) International publication number: WO 2004/060801 (22.07.2004 Gazette 2004/30)
- (84) Designated Contracting States:

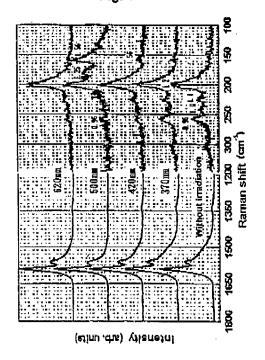
 AT BE BG CH CY CZ DE DK EE ES FI FR GB GR
 HU IE IT LI LU MC NL PT RO SE SI SK TR
- (30) Priority: 06.01.2003 JP 2003000718
- (71) Applicants:
 - Japan Science and Technology Agency Kawaguchi-shi, Saitama 332-0012 (JP)
 - NEC CORPORATION
 Minato-ku, Tokyo 108-8001 (JP)

- (72) Inventors:
 - IJIMA, Sumio
 Nagoya-shi, Alchi 468-0011 (JP)
 - YUDAŞAKA, Masako Tsukuba-shi, ibaraki 300-2635 (JP)
 - ZHANG, Minfang
 Tsukuba-shi, Ibaraki 305-0861 (JP)
- (74) Representative: Calamita, Roberto Frank B. Dehn & Co.,
 European Patent Attorneys,
 179 Queen Victoria Street
 London EC4V 4EL (QB)

(54) METHOD OF SELECTING STRUCTURE OF CARBON NANOTUBE THROUGH PHOTOIRRADIATION

(57) A method for selectively obtaining carbon nanotubes having specific structures by selectively obtaining carbon nanotubes having structures different from the structures of the carbon nanotubes to be vanished by irradiating carbon nanotubes with a light beam of single wavelength so as to have carbon nanotubes in specific electron states in the excited states, and oxidizing and combusting the carbon nanotubes in the excited states by an oxygen or an oxidizing agent so as to vanish the same.

Fig. 1



Printed by Jouve, 75001 FARIS (FR)

2

Description

Technical Field

[0001] The present invention relates to a method for selecting structures for carbon nanotubes by the light irradiation. More specifically, the present invention relates to a method for selecting structures for carbon nanotubes by the light irradiation, capable of selectively obtaining only carbon nanotubes having specific structures by the light irradiation and the combustion at a low temperature.

Background Art

[0002] Since the discovery of the carbon nanotube, its application to electronic semiconductors, electronic devices, or other fields has been strongly expected, and various studies have been carried out by a large number of researchers.

[0003] Among the carbon nanotubes, for example, a single-wall carbon nanotube has a shape with a piece of a graphite sheet comprising a six membered ring of carbons rolled like a cylinder. Depending on the rolling style of the graphite sheet, that is, the diameter of the carbon nanotube and the chirality (helical degree), the conductivity of the carbon nanotube differs completely so as to provide a metal or a semiconductor.

[0004] However, according to the production method for carbon nanotubes so far, the diameters or the chiralities of the carbon nanotubes to be produced cannot be controlled so that only inhomogeneous ones can be obtained, and thus the difference of the conductivity of the carbon nanotubes provided by the diameters or the chiralities has not been utilized sufficiently.

[0005] In contrast, recently, it was revealed that the chemical reaction can be promoted by the light irradiation. These days, it was also found out that the Flahren's reaction is carried out by the assistance of the light exoltation. In view of these facts, it is considered that the light irradiation has a possibility to have any influence on the chemical reaction of the single-wall carbon nanotubes (SWNTs), however, the influence to the chemical reaction of the single-wall carbon nanotubes by the light irradiation has been completely unknown so far.

[0006] Then, in view of the above-mentioned circumstances, the present invention has been achieved, and its object is to provide a method for selectively obtaining carbon nanotubes, capable of solving the problems of the conventional technique, having structures different from the structures of the carbon nanotubes to be vanished by selectively combusting and vanishing carbon nanotubes of specific structures.

Disclosure of invention

[0007] In order to achieve the above-mentioned object, the present invention firstly provides a method for

selecting structures for carbon nanotubes by the light Irradiation, characterized in selectively obtaining carbon nanotubes having structures different from the structures of the carbon nanotubes to be vanished by irradiating carbon nanotubes with a light beam of single wavelength so as to have carbon nanotubes in a specific electron states in the excited states, and oxidizing and combusting the carbon nanotubes in the excited states by an oxygen or an oxidizing agent so as to vanish the same.

[0008] The present invention secondly provides the method for selecting structures for carbon nanotubes by the light irradiation according to the first invention, characterized in that the carbon nanotubes in the excited states are vanished by oxidizing and combusting at a temperature of 0°C or more and 500°C or less.

[0009] It thirdly provides the method for selecting structures for carbon nanotubes by the light irradiation according to the first or second invention, characterized in that the oxidizing agent is a hydrogen peroxide water, a nitric acid or a potassium permanganate.

[0010] It fourthly provides the method for selecting structures for carbon nanotubes by the light irradiation according to any one of the first to third inventions, characterized in that light beams having different wavelengths are irradiated to the carbon nanotubes respectively for selectively oxidizing and combusting carbon nanotubes having specific structures corresponding to the wavelength of each light beam so as to vanish the same.

[0011] Furthermore, it fifthly provides the method for selecting structures for carbon nanotubes by the light irradiation according to any one of the first to third inventions, characterized in that only carbon nanotubes having specific structures are selectively obtained by irradiating the carbon nanotubes successively with a plurality of light beams having different wavelengths.

[0012] Moreover, it sixthly provides the method for selecting structures for carbon nanotubes by the light irradiation according to any one of the first to fifth invontions, characterized in that the carbon nanotubes are single-wall carbon nanotubes.

Brief Description of Drawings

[0013]

45

50

FIG 1 is a graph showing an example of Raman spectra of an embodiment of the present invention. FIG 2 is a graph showing an example of absorption spectra of an embodiment of the present invention. FIG 3 is a graph showing an example of Raman spectra of another embodiment of the present invention.

Best Mode for Carrying Out the Invention

[0014] The present invention has the above-men-

3

EP 1 591 417 A1

tioned characteristics, and the embodiments thereof will be explained hereafter.

[0015] A method for selecting structures for carbon hanotubes by the light irradiation according to the present invention is largely characterized in selectively obtaining only carbon nanotubes having structures different from the structures of the carbon nanotubes to be vanished by irradiating carbon nanotubes with a light beam of single wavelength so as to have carbon nanotubes in specific electron states in the excited states by absorbing the light beam, and oxidizing and combusting the carbon nanotubes in the excited states by an oxygen or an oxidizing agent so as to vanish the same. At the time, the carbon nanotubes in the excited states can be vanished by preferably oxidizing and combusting at a temperature of 0°C or more and 500°C or less.

[0016] That is, by irradiating the carbon nanotubes with a light beam of a single wavelength, only carbon nanotubes having specific structures can be vanished by combustion at a low temperature of 0°C or more and 500°C or less, and as a result, carbon nanotubes having structures different from the carbon nanotubes to be vanished can be obtained selectively.

[0017] As the light beam to be irradiated to the carbon nanotubes, any light beam can be used as long as It is a single wavelength light beam, and it may be a laser light beam or a non laser light beam. Moreover, in the case the carbon nanotubes are to be oxidized by an oxygen, It may be carried out in the atmosphere with the presence of an oxygen to the degree capable of oxidizing the carbon nanotubes. For example, It may be in the atmosphere including not only the oxygen, such as in the air. In the atmosphere including the oxygen, carbon nanotubes of specific structures can be combusted in a temperature range of 100°C to 500°C.

[0018] Moreover, on the other hand, as an oxidizing agent for oxidizing the carbon nanotubes, an optional oxidizing agent can be used. In particular, a hydrogen peroxide water, a nitric acid or a potassium permanganate can be used preforably. For example, in the case a hydrogen peroxide water (concentration 10 to 30%) is used, carbon nanotubes of specific structures can be combusted and vanished in a range of 0°C to 100°C.

[0019] As mentioned above, according to the method for selecting structures for carbon nanotubes by the light irradiation of the present invention, since carbon nanotubes having specific structures can be excited for promoting the oxidization thereof by irradiating the carbon nanotubes with a single wavelength light beam, and thereby the carbon nanotubes excited by heating at a low temperature can be combusted and vanished, carbon nanotubes of a good quality, having specific structures can be obtained without damaging the carbon nanotubes remaining without vanishing.

[0020] Moreover, according to the method for selecting structures for carbon nanotubes by the light irradiation of the present invention, since light beams of different wavelengths are irradiated to the carbon nanotubes

for selectively oxidizing and combusting carbon nanotubes having specific structures corresponding to the wavelength of each light beam so as to vanish the same, carbon nanotubes of necessary structures can optionally be obtained selectively.

[0021] Furthermore, since a plurality of light beams having different wavelengths are irradiated to the carbon nanotubes successively, only carbon nanotubes having specific structures further limited compared with the case of irradiating with a light beam of a light beam of one wavelength can be obtained selectively.

[0022] By using the above-mentioned method, carbon nanotubes having the necessary electric characteristics such as a metal and a semiconductor can selectively be obtained easily.

[0023] The method for selecting structures for carbon nanotubes by the light irradiation according to the present invention can be carried out particularly for single-wall carbon nanotubes so that single-wall carbon nanotubes having specific structures with the necessary electric characteristics can easily be obtained certainly. [0024] Hereinafter, with reference to the accompanied drawings, the embodiments of the present invention will be explained in further details. Of course the invention is not limited to the examples herebelow, and it is needless to say that various embodiments can be used for the details.

Examples

<Example 1>

[0025] Single-wall carbon nanotubes produced by the HIPco method as a method for synthesizing carbon nanotubes by heating a carbon monoxide at a high pressure (CO + CO \rightarrow C + CO₂) were treated with HCl for eliminating Fe. Then, light beams having 370 nm, 420 nm, 500 nm and 620 nm wavelengths were used for the light irradiation each at 320°C for 2 hours in the air.

[0026] The Raman spectra thereof are shown in FIG. 1. As it is known by the comparison with the case without the light irradiation, although a light beam having a 370 nm wavelength does not influence the oxidization of the single-wall carbon nanotubes at all, the light beams of the other wavelengths promoted the oxidization of the single-wall carbon hanotubes having specific structures. In the case a light beam of a 420 nm wavelength is irradiated to the single-wall carbon nanotubes, as it is shown by the Raman spectrum, the single-wall carbon nanotubes of a 0.96 nm diameter and a 1.0 nm diameter were completely vanished, and those of 1.1 nm and 1.2 nm diameters remained without vanishing. In the case of the Irradiation with a light beam of a 500 nm wavelength, although the single-wall carbon nanotubes of about 1.0 nm and about 1.1 nm diameters were vanished, on the other hand in this case, two novel singlewall carbon nanotubes of about 1.35 nm and about 1.56 nm diameters appeared.

5

EP 1 591 417 A1

[0027] Furthermore, as a result of executing the light Irradiation using a light beam of a 620 nm wavelength, only single-wall carbon nanotubes of about a 1.2 nm diameter remained, and the carbon nanotubes of the other diameters were vanished. It was shown that the substantially the same results were provided for the oxygen concentration and the kind of the chemical bond of C and O regardless of the light Irradiation of the different wavelengths or without the light irradiation in the all specimens formed by the HiPco method from the X ray photo electron spectroscopy (XPS). Thereby, it was learned that the vanished single-wall carbon nanotubes were combusted selectively instead of the chemical reaction with the oxygen or the formation of a carbonyl or a carboxyl compound.

[0028] Next, FIG 2 shows absorption spectra of the case without the light irradiation and the cases of the light irradiation using light beams of 370 nm, 420 nm, 500 nm, and 620 nm wavelengths each at 320°C for 30 minutes. Although the absorption spectra shows that the light irradiation does not influence the band S_{11} peak (corresponding to the first intermediate transition), the S_{22} (corresponding to the second intermediate transition) peak intensity of 1.37 eV corresponds to singlewall carbon nanotubes of a 1.2 nm diameter, with the intensity increased after the light irradiation of 420 nm, 500 nm or 620 nm, and the S_{22} peak intensity of 1.5 eV and 1.63 eV (corresponding to the single-wall carbon nanotubes of 1.1 nm and 1.0 nm diameters) were reduced.

[0029] These results show that the light irradiation promoted the oxidization of the single-wall carbon nanotubes, and that a light beam of a specific wavelength selectively oxidized, combusted and vanished single-wall carbon nanotubes of specific structures.

<Example 2>

[0030] Next, FiG 3 shows the Raman spectra at the time of placing single-wall carbon nanotubes produced by the same method as in the example 1 in a hydrogen peroxide water, and executing the light Irradiation for 2 minutes using a light beam having a 488 nm wavelength in the hydrogen peroxide water of 100°C. For the comparison, FiG 3 also shows the Raman spectrum in the case without the light irradiation in states with the single-wall carbon nanotubes placed in a hydrogen peroxide water of 100°C.

[0031] As it is apparent from FIG 3, in the case of the light irradiation of a 488 nm wavelength, the peak at 200 cm⁻¹ is reduced, and in contrast, in the case without the light irradiation, the Raman spectrum change was not observed before and after the process with the hydrogen peroxide water.

[0032] Therefore, the experiment results show that 55 the light irradiation promotes the oxidization of the single-wall carbon nanotubes also in the case of using an oxidizing agent such as a hydrogen peroxide water, and

It denotes that a light beam of a specific wavelength setectively oxidizes, combusts and vanishes single-wall carbon nanotubes of specific structures.

6

5 Industrial Applicability

[0033] As heretofore explained in detail, according to the present invention, a method for selecting structures for carbon nanotubes by the light irradiation, capable of selectively obtaining only carbon nanotubes having speoific structures by the light irradiation and the combustion at a low temperature can be provided.

16 Claims

35

- A method for selecting structures for carbon nanotubes by the light irradiation, characterized in selectively obtaining carbon nanotubes having structures different from the structures of the carbon nanotubes to be vanished by irradiating carbon nanotubes with a light beam of a single wavelength so as to have carbon nanotubes in specific electron states in the excited states, and oxidizing and combusting the carbon nanotubes in the excited states by an oxygen or an oxidizing agent so as to vanish the same.
- 2. The method for selecting structures for carbon nanotubes by the light irradiation according to claim 1, characterized in that the carbon nanotubes in the excited states are vanished by oxidizing and combusting at a temperature of 0°C or more and 500°C or less.
 - The method for selecting structures for carbon nanotubes by the light irradiation according to claim 1 or 2, characterized in that the oxidizing agent is a hydrogen peroxide water, a nitric acid or a potassium permanganate.
 - 4. The method for selecting structures for carbon nanotubes by the light irradiation according to any one of claims 1 to 3, characterized in that light beams having different wavelengths are irradiated to the carbon nanotubes respectively for selectively oxidizing, combusting and vanishing carbon nanotubes having specific structures corresponding to the wavelength of each light beam.
 - 5. The method for selecting structures for carbon nanotubes by the light irradiation according to any one of claims 1 to 3, characterized in that only carbon nanotubes having specific structures are selectively obtained by irradiating the carbon nanotubes successively with a plurality of light beams having different wavelengths.

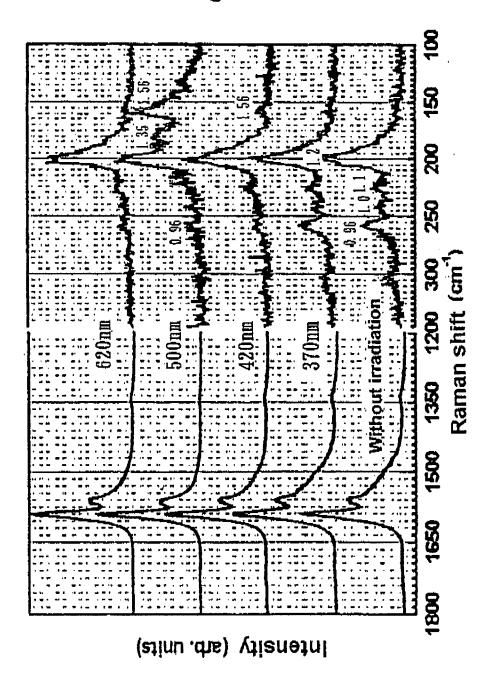
50

EP 1 591 417 A1

The method for selecting structures for carbon nanotubes by the light irradiation according to any one of claims 1 to 5, characterized in that the carbon nanotubes are single-wall carbon nanotubes.

5

Fig. 1



e

Fig. 2

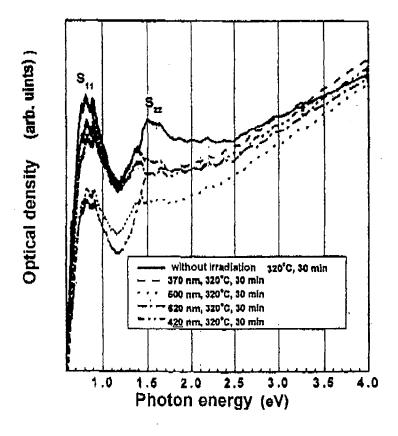
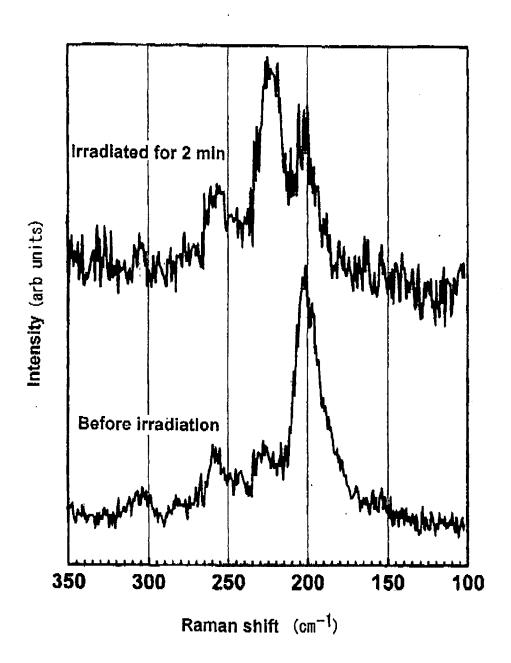


Fig. 3



INTERNATIONAL SEARCH REPORT			International application No.			
			PCT/JP	03/04182		
A. CLASSIFICATION OF SÜBJECT MATTER Int.Cl' C01B31/02						
According to International Patent Classification (IPC) or to both national classification and IPC						
B. FIELDS SEARCHED						
Minimum documentation searched (classification system followed by classification symbols) Int.Cl ² CO1B31/02						
Documentation searched other than minimum documentation to the extent that such documents are included in the fields searched						
Jitsuyo Shinan Koho 1926-1996 Jitsuyo Shinan Toroku Koho 1996-2003						
Electronic data base consulted during the international search (name of data base and, where practicable, search terms used) JSTPlus FILE (JOIS), Web of Science, Science Direct						
!						
C. DOCUMENTS CONSIDERED TO BE RELEVANT						
Category*	Citation of document, with indication, where ap	propriate, of the relev	unt passages	Relevant to claim No.		
A	P.D.KICHAMBARE et al., Laser			1-6		
	nanotubes, MATERIALS CHEMISTS Vol.72, pages 218 to 222	(Y AND PHISIC	38, 2001,			
A	N.BRAIDY et al., Oxidation of	Fe Nanopart	icles:	1-6		
	Embedded in Single-Walled Car Exposure to a Bright Flash of	bon Nanotube	ea by	- •		
	LETTERS, 2002, Vol.2, No.11,	pages 1277 t	o 1280			
A	WO 98/39250 A1 (WILLIAM MARS		RSITY),	1-6		
	11 September, 1998 (11.09.98) Full text	11 September, 1998 (11.09.98),				
·	& AU 9865456 A & EP 1015384 A1 & JP 2002-515847 A & US 2002/085968 A1					
,	« OF 2002-31304/ A	2002/083968	ΑI			
(Table 1)						
Further documents are listed in the continuation of Box C. See patent family annex.						
Special categories of cited documents: "A" document defining the general state of the art which it not considered to be of particular relevance understand the original of the original filting date or priority date and not in conflict with the application but clear to understand the principle or theory underlying the investion						
"E" earlier document but published on or after the international filting "X" document of particular rejovance; the claimed invention current by						
"L" ducument which may throw doubts on priority claim(s) or which is step when the document is taken slone ofted to establish the publication dute of snotker citation or other "Y" document of particular relevance; the claimed invention cannot in the company of particular relevance; the claimed invention cannot in the company of the company						
special recision (at specified) considered to involve an inventive step when the document is combined with one or more other such documents, such combined with one or more other such documents, such						
mouse "P" document published prior to the informational filing date but later than the priority date dulmed "E" document published prior to the informational filing date but later than the priority date dulmed "E" combination being obvious to a person skilled in the art document member of the same patent family						
Date of moiting of the international search 17 June, 2003 (17.06.03) Date of moiting of the international search report 01 July, 2003 (01.07.03)						
01 0HLY, 2005 (01.07.03)						
	alling address of the ISAV ness Patent Office	Authorized officer				
Pacrimite No.		Telephone No.				

Form PCT/ISA/210 (second sheet) (July 1998)

International application No. INTERNATIONAL SEARCH REPORT

		PCT/JP03/04182		
C (Continuation). DOCUMENTS CONSIDERED TO BE RELEVANT Catagory* Citation of document, with indication, where appropriate, of the relevant passages Relevant to claim				
A	JP 2002-255519 A (Toyota Central Research And Development Laboratories, Inc.), 11 September, 2002 (11.09.02), Full text (Family: none)			
Α.	JP 2002-80211 A {Canon Inc.}, 19 March, 2002 (19.03.02), Full text (Family: none)	1-6		
A	EP 1129990 A1 (LUCENT TECHNOLOGIES INC.), 05 September, 2001 (05.09.01), & AU 200123085 A & JP 2001-262343 A & CA 2331278 A1 & KR 200185509 A	. 1-6		
E,A	JP 2003-146631 A (Japan Science and Technology Corp.), 21 May, 2003 (21.05.03), Full text (Family; none)	y 1-6		
T	M.YUDASAKA et al., Diameter-selective removal of single-wall carbon nanotubes through light-assisted oxidation, CHEMICAL PHYSICS LETTERS, 04 June, 2003 (04.06.03), Vol.374, pages 132 to 136	1-6		
		,		

Form PCT/ISA/210 (continuation of second sheet) (July 1998)